

This article was downloaded by:

On: 26 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713597286>

### Synthesis and Anti-HIV Evaluation of New 5-Substituted-2',3'-thiauridine Nucleosides

Nicolas Mourier<sup>ab</sup>; Carole Trabaud<sup>ab</sup>; Valerie Niddam<sup>ab</sup>; Jean-Christophe Graciet<sup>ab</sup>; Michel Camplo<sup>ab</sup>; Jean-Claude Chermann<sup>b</sup>; Jean-Louis Kraus<sup>ab</sup>

<sup>a</sup> Laboratoire de Chimie Biomoléculaire, Faculté des Sciences de Luminy, MARSEILLE, FRANCE <sup>b</sup> INSERM Unité U-322, Unité des Rétrovirus et Maladies associées, Campus Universitaire de Luminy, MARSEILLE, FRANCE

**To cite this Article** Mourier, Nicolas , Trabaud, Carole , Niddam, Valerie , Graciet, Jean-Christophe , Camplo, Michel , Chermann, Jean-Claude and Kraus, Jean-Louis(1996) 'Synthesis and Anti-HIV Evaluation of New 5-Substituted-2',3'-Dideoxy-3'-thiauridine Nucleosides', *Nucleosides, Nucleotides and Nucleic Acids*, 15: 7, 1397 — 1409

**To link to this Article:** DOI: 10.1080/07328319608002439

**URL:** <http://dx.doi.org/10.1080/07328319608002439>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

## Synthesis and anti-HIV evaluation of new 5-substituted-2',3'-dideoxy-3'-thiauridine nucleosides

Nicolas Mourier<sup>1,2</sup>, Carole Trabaud<sup>1,2</sup>, Valerie Niddam<sup>1,2</sup>, Jean-Christophe Graciet<sup>1,2</sup>, Michel Camplo<sup>1,2</sup>, Jean-Claude Chermann<sup>2</sup> and Jean-Louis Kraus<sup>1,2\*</sup>.

- 1 Laboratoire de Chimie Biomoléculaire, Faculté des Sciences de Luminy, case 901, 13288 MARSEILLE CEDEX 9, FRANCE.
- 2 INSERM Unité U-322, Unité des Rétrovirus et Maladies associées, Campus Universitaire de Luminy. BP 33, 13273 MARSEILLE CEDEX 9, FRANCE.  
Fax (33) 91.41.92.50.

**Abstract:** On the basis of molecular modeling calculations using GenMol software, new 5-substituted-2',3'-dideoxy-3'-thiauridine were designed as possible anti-HIV reverse transcriptase inhibitors. The synthesis of the key intermediate 5-carboxy-2',3'-dideoxy-3'-thiauridine was achieved through the condensation of the fully silylated 5-carboxyuracil on 2-benzoyl methyl-5-acetoxy-1,3-oxathiolane using trimethylsilyl triflate (TMSOTf). This latter compound was condensed with 2-(*N*-*tert*-Butoxycarbonyl)-1-aminoethane in the presence of *N,N*-diisopropylethylamine (DIEA). The subsequent carboxamide deprotection led to the final compounds. These new analogues were evaluated for their anti-HIV-1 activities on infected MT<sub>4</sub> cells but no significant protection was observed. Electronic and structural parameters considered in this model were not sufficient to predict any active anti-HIV molecular structures.

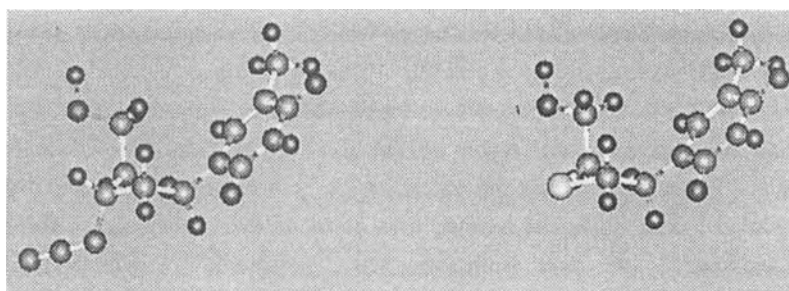
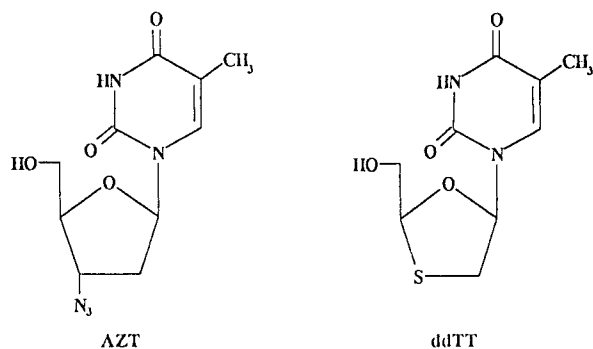
## INTRODUCTION

Antiretroviral agents developed today have been primarily the inhibitors of HIV-reverse transcriptase. The HIV RT has been one of the most intensively studied viral targets for the development of anti-HIV drugs. 2',3'-dideoxy analogues can be metabolized to form potent RT. Chain terminator (1) have been identified to elicit potent antiretroviral activity.

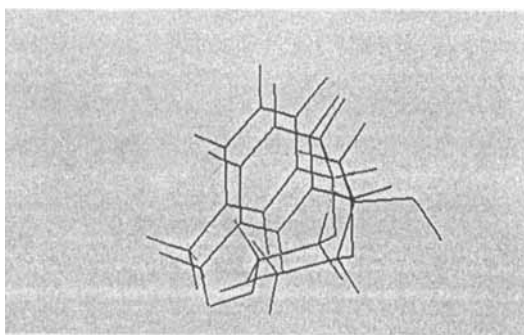
Dideoxynucleosides are of special interest since "simple" chemical modification of their sugar moiety can convert a normal substrate into a potent inhibitor for HIV replication. Analysis of the molecular structure of the most active anti-HIV nucleosides [AZT (Retrovir), ddI (Videx), ddC (Zalcitabine), d4T (Stavudine) and 3TC (Lamivudine)] reveals that for a given ribose ring only a few specific nucleic bases led to active anti viral drugs. Having actively participated to the discovery of 2',3'-dideoxy-3'-thiacytidine (3TC) (1,2,3) and related drugs (4,5,6,7), we have investigated the possibilities to use the promising heterocycle 1,3-oxathiolane as a ribose ring mimic coupled with modified uracil or thymine nucleic base. Molecular modeling calculations using GenMol software (5) were undertaken in order to evaluate the energetical differences between the lowest energy conformer of 2',3'-dideoxy-3'-azidothymidine (AZT) and its corresponding 2',3'-dideoxy-3'-thiathymidine (ddTT). From these calculations, we synthesized new 5-substituted 2',3'-dideoxy-3'-thiauridine, and proceed to their antiviral evaluation.

### Conformational studies

All the molecules with modified or unmodified ribose are considered in the C3' exo conformation as deduced from X-Ray analysis by Van Roey and al. (2). According to these authors, C3' exo conformation was the one active on the kinase and RT receptors. In order to have a chance to fit into the kinase or RT active site, the nucleoside drug must have an envelope (related to conformation) in terms of both geometry and electrostatic potential equivalent to that of the natural substrate. This is required to be complementary to the active site envelope (3). Therefore necessary energy to put the new 5-substituted-2',3'-dideoxythiauridine into cytidine-like geometry was calculated. The difference between strain energy of an analogue in the cytidine-like geometry and the minimum strain energy of this analogue in C3' conformation, corresponds to the increase of the transition state energy with cytidine. Its effect on the kinetic constant of the fit was calculated. These calculations based on Force Field and MEP (Molecular Electrostatic Potential) (4), were performed with GenMol software (5). GenMol is a  $3.10^5$  instructions software, with an original Force Field able to treat atomic and molecular systems containing up to  $10^5$  atoms of 96 different types. The lowest energy conformers of AZT, 2',3'-dideoxy-3'-thiathymidine (ddTT) and 5-substituted-2',3'-dideoxythiauridine derivatives were examined and characterized. As shown on Fig.1, the lowest energy conformation of both AZT and ddTT was found to be very similar. Force Field and Molecular Electrostatic Potential (MEP) calculations indicated in the case of AZT that a 3'-electron withdrawing substituent N3 induced a slightly positive MEP. ddTT with a sulfur atom in 3' position displayed also a weak negative MEP in the same region. In contrast MEP effect disappeared in the case of the corresponding 3'-deoxyribose thymidine. Since both AZT and Thymidine fits the reverse transcriptase active site, this result could indicate that the active site has a weak MEP selectivity. *In vitro* anti-HIV experiments on infected MT<sub>4</sub> cells showed that ddTT was inactive (8), while AZT was one of the most active anti-HIV drug. This

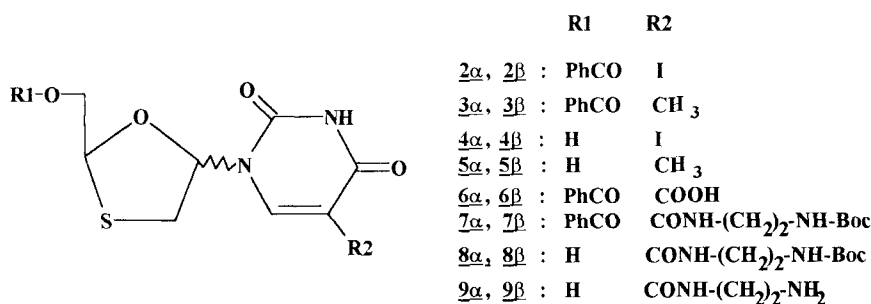


**1a** Stereoviews of AZT (left) and ddTT (2',3'-dideoxy-3'-thiathymidine, right) in their lowest conformation energy.



**1b** Overlay of AZT and ddTT as found in the surimposed structure determinations in their HIV-RT bound conformations.

**Fig. 1-** Lowest energy conformation of AZT and its 3'-oxathiolane derivative.

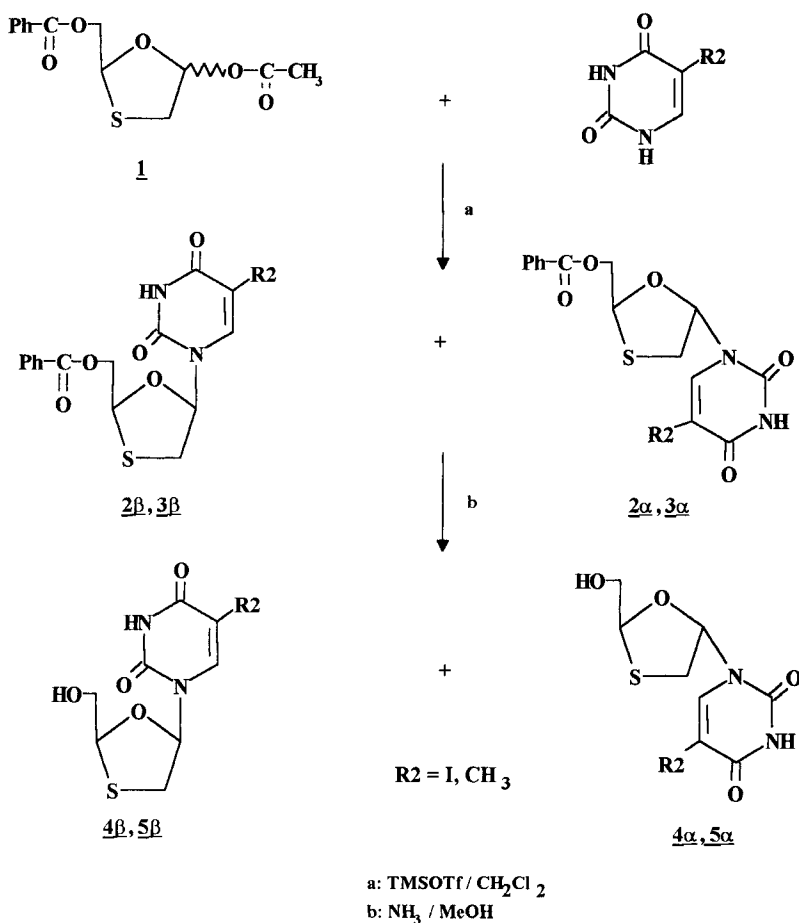
**Fig 2**

biological observation indicates that parameters others than conformation energy, or geometrical effects are involved in the mechanism which confer an antiretroviral activity. Anti-HIV activity of dideoxynucleosides is critically dependent of triphosphorylation mediated by cellular kinase (9,10,11), but also depends on the bioavailability of the drug inside the infected cells. Taking into account results report by Van Roey (2) that the C3' exo conformation deduced from X-Ray analysis was the one active on both kinase and RT active site, then biodelivery of the drug inside the infected cells could be the limiting factor for anti-HIV activity. Consequently, we have synthesized new 5-substituted-2',3'-dideoxy-3'-thiauridine which were submitted to anti-HIV evaluation. The 5-substituents were selected for their large difference in polarity (COOH, CONH-(CH<sub>2</sub>)<sub>2</sub>-NH-Boc, CONH-(CH<sub>2</sub>)<sub>2</sub>-NH<sub>2</sub>, CH<sub>3</sub>, I) which can confer to the resulting molecules different lipophilic characters.

## Chemistry

The following new 2',3'-dideoxy-3'-thia C-5 substituted uracilnucleosides were synthesised (Fig 2).

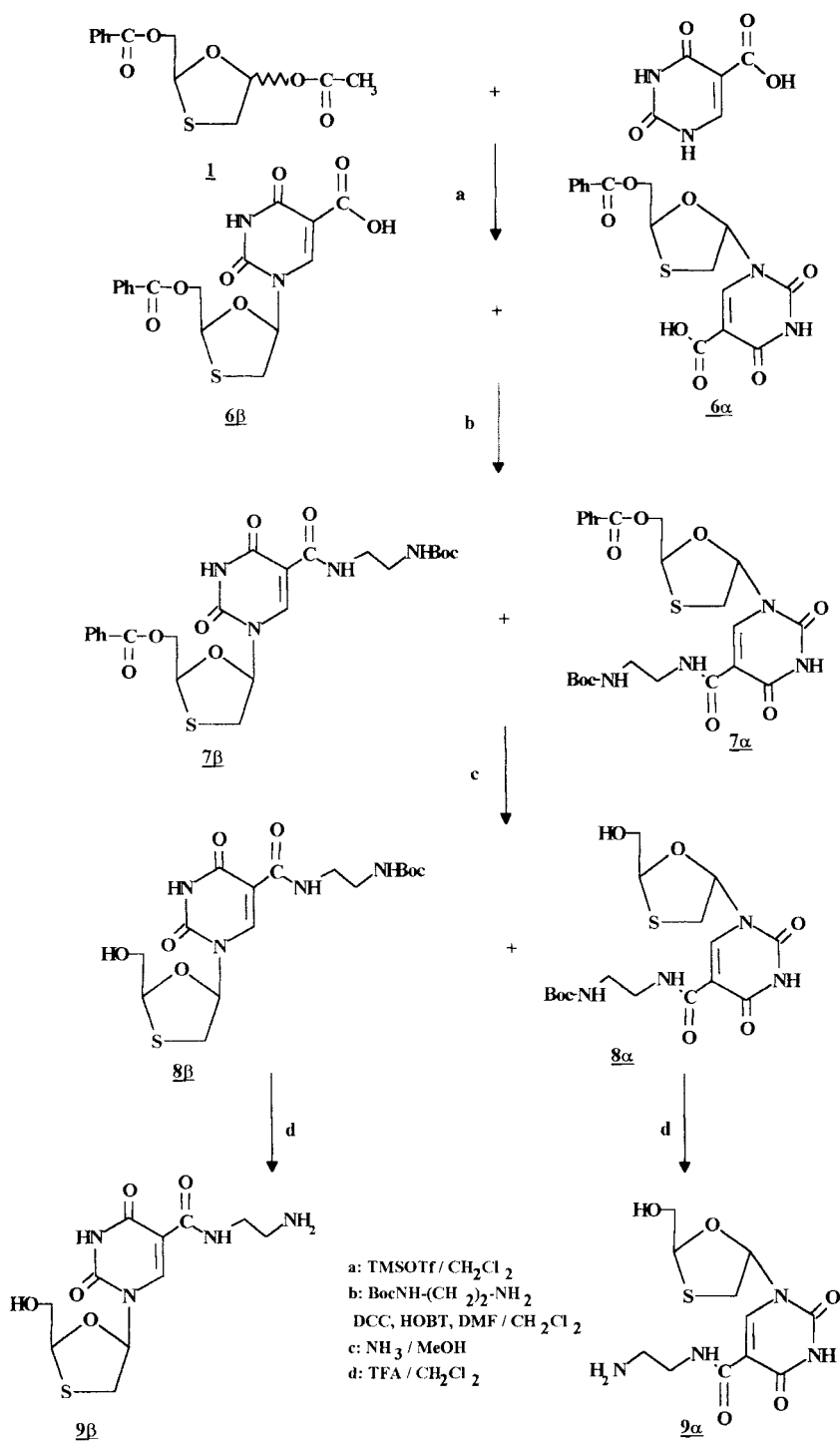
The synthesis of these new oxathiolane nucleosides is illustrated on scheme 1 and scheme 2. The key intermediate 1,3-oxathiolane (**1**) was synthesised according to standard procedures already reported (12,13,14). In an attempt to find the optimal conditions for the condensation of 2-benzoyloxymethyl-5-acetoxy-1,3-oxathiolane (**1**) with the silylated 5-carboxy uracil base, various Lewis acid (TMSOTf, SnCl<sub>4</sub> and TiCl<sub>4</sub>), solvents (CH<sub>3</sub>CN, CH<sub>2</sub>Cl<sub>2</sub>, ClCH<sub>2</sub>CH<sub>2</sub>Cl) and reaction thermodynamic conditions were investigated. We were finally able to synthesize the corresponding  $\alpha$  and  $\beta$  forms of 5-substituted uracil derivatives (**6** $\alpha$  and **6** $\beta$ ) using TMSOTf in dichloromethane at room temperature for 8h. The 1:1 mixture of  $\alpha$  and  $\beta$  isomers was separated using column chromatography (eluent: CH<sub>2</sub>Cl<sub>2</sub>/ CH<sub>3</sub>OH 9:1) with an overall yield of 90%. The next step involved the coupling between a mono-protected  $\omega$ -diamine and the 5-carboxylic function. As a preliminary assay, aminoethylcarboxamide prepared according to a described procedure (15) was coupled to **7** $\alpha$  and **7** $\beta$ . Several coupling

**Scheme 1**

conditions were investigated:

- BOP (benzotriazol-1-yloxy-tris (dimethylamino) phosphonium-hexafluoro-phosphate) in DMF (21,22,4) in the presence of TEA.
- DCC (4-dicyclohexylcarbodiimide), HOBT (hydroxybenzotriazole) in DMF, and DCC/HOBT in the presence of DIEA.

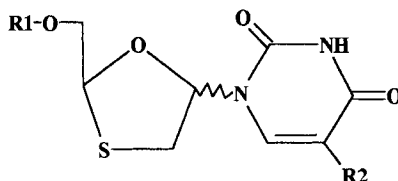
We found that the best yields (65%) were obtained using DCC/HOBT in  $\text{CH}_2\text{Cl}_2$  with DIEA. Deprotection of compounds 7 $\alpha$  and 7 $\beta$  using saturated methanolic ammonia solution led to the corresponding analogues 8 $\alpha$  and 8 $\beta$  in a quantitative yield. In order to avoid oxathiolane ring opening, very mild acidic conditions (98% formic acid or trifluoroacetic acid) were used at room temperature to remove Boc aminoprotecting group (16,17). Final compounds 9 $\alpha$  and 9 $\beta$  were obtained in 68% and 98% yield, respectively. The determination



Scheme 2

**Table 1**

Anti-HIV-1 activity of 5-substituted 2',3'-dideoxy-3'-thiacytidine.



N°	R1	R2	Stereochemistry	IC <sub>50</sub> *μM
<b>4</b> $\alpha$	H	I	$\alpha$	>100
<b>4</b> $\beta$	H	I	$\beta$	>100
<b>5</b> $\alpha$	H	CH <sub>3</sub>	$\alpha$	>200
<b>5</b> $\beta$	H	CH <sub>3</sub>	$\beta$	>200
<b>7</b> $\alpha$	PhCOO	CONH(CH <sub>2</sub> ) <sub>2</sub> NHBoc	$\alpha$	>200
<b>7</b> $\beta$	PhCOO	CONH(CH <sub>2</sub> ) <sub>2</sub> NHBoc	$\beta$	>200
<b>9</b> $\alpha$	H	CONH(CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub>	$\alpha$	>200
<b>9</b> $\beta$	H	CONH(CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub>	$\beta$	>200
AZT	-	-	$\beta$	0.05
$\pm$ 3TC	H	H	$\beta$	0.1

\* IC<sub>50</sub> : concentration required to produce 50% inhibition of syncytia formation on MT<sub>4</sub> cells.

of the anomeric configuration of compounds **9** $\alpha$  and **9** $\beta$  was confirmed by NMR NOESY experiments. Structural assignments of the compounds described in the experimental part were based on elemental analysis, <sup>1</sup>H, <sup>13</sup>C NMR and mass spectra data. <sup>13</sup>C NMR signal assignments were confirmed by DEPT-135 experiments.

#### Anti-HIV activity and discussion

The potency of the new synthesised 2',3'-dideoxy-5-substituted-3'-thiauridine as inhibitors of HIV-replication was evaluated. The anti-HIV-1 activity was determined through the formation of syncytia (18,19) in infected MT<sub>4</sub> cells (20). IC<sub>50</sub> values (concentration of drugs required to



produce 50% inhibition of the syncytia formation) are reported in Table 1. The first observation which can be inferred from these results is that none of the new derivatives in their  $\alpha$  or  $\beta$  forms elicited anti-HIV activity. Moreover, as already reported (8), ddTT (5 $\beta$ ) which represents the corresponding analogue of ( $\pm$ ) 3TC is denied of anti-HIV activity. The observed anti-HIV inactivity of these 5-substituted 2',3'-dideoxy-3'-thiauridine derivatives suggest the following comments. Substitution at the C5 position of the uracil base by chemical groups having different lipophilic characters did not confer to the resulting molecule any anti-HIV activity. Therefore, the lowest energy conformation C3' exo, selected by the modeling molecular calculations to be the one active on both kinase and RT enzyme active sites is not suitable in the case of new 5-substituted 2',3'-dideoxy-3'-thiauridine. This result seems to indicate a greater sensitivity of the thymine kinases to the thymidine structure modifications.

In conclusion, the synthesis of new 5-substituted-2',3'-dideoxy-3'-thiauridine derivatives has been achieved. These compounds were designed in order to elicit anti-HIV properties. The introduction of heteroatoms like sulfur in the backbone sugar ring could lead to active anti-HIV nucleosides, (3TC). At the opposite the replacement of cytosine of 2',3'-dideoxy-3'-thiacytidine by different uracils substituted at C-5 by groups having different lipophilic characters seems to abolish the anti-HIV properties.

## EXPERIMENTAL SECTION

Proton magnetic spectra were recorded on a Bruker AMX 200 spectrometer. Chemical shifts were reported as values in parts per million. Coupling constants were expressed in hertz (Hz). Elemental microanalysis were determined by Service Central d'Analyse CNRS Vernaison-Lyon France and gave combustion values for C, H, N within 0.4% of theoretical values. Analytical thin layer chromatographies (TLC) were carried out on aluminium precoated TLC plates with silica gel Kieselgel 60 F<sub>254nm</sub> 0.2 mm thickness (Merck and Co, Darmstadt). Column flash chromatographies were performed with Merck silica gel (230-400 mesh). Preparative layer chromatographies were carried out on silica gel 60 F<sub>254nm</sub> precoated PLC plates (20 x 20 cm layer thickness 1 or 2 mm). Melting points were determined with a MEL-TEMP capillary apparatus. FAB<sup>+</sup> mass spectra were recorded on a JEOL DX-100 mass spectrometer at the Laboratoire de Mesures Physiques-RMN, USTL, Montpellier, France.

### Chemistry

2',3'-dideoxy-3'-thiathymidine  $\alpha$  and  $\beta$  (5 $\alpha$  and 5 $\beta$ ) forms have been synthesized according to a procedure described in reference 13.

#### 2',3'-dideoxy-5'-benzoyloxymethyl-5-iodo-3'-thiauridine 2 $\alpha$ and 2 $\beta$ :

A solution of silylated 5-iodouracil (1eq, 1.7 g, 3.6 mmol) in dry dichloromethane (10 mL) was added to a solution of compound 1 (1eq, 1.0 g, 3.6 mmol) in dry dichloromethane (10 mL).

The mixture treated with trimethylsilyl triflate became clear after stirring for 3h, under nitrogen atmosphere at room temperature. It was then evaporated under reduced pressure. An aqueous solution (5%)  $\text{NaHCO}_3$  was added and the product was extracted with dichloromethane. The organic phases were washed with  $\text{H}_2\text{O}$  and dried over  $\text{Na}_2\text{SO}_4$  to give after solvent evaporation a white solid (1.514 g). Yield (91%). TLC ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$  95:5)  $R_f$  = 0.40.

**2 $\alpha$  isomer:**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 3.13-3.30 (m, 1H, H-2'a or H-2'b); 3.54-3.71 (m, 1H, H-2'a or H-2'b); 4.41-4.60 (m, 1H, H-5'a or H-5'b); 4.70-4.77 (m, 1H, H-5'a or H-5'b); 5.52 (t, 1H,  $J$  = 3.85 Hz, H-4'); 6.34 (t, 1H,  $J$  = 5.10 Hz, H-1'); 7.47-8.15 (m, 6H, H arom.); 9.29 (s br, 1H, NH).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 37.9 (C-2'); 66.7 (C-5); 69.4 (C-5'); 84.8 (C-4'); 88.5 (C-1'); 129.0-134.0 (C arom.); 144.4 (C-6); 150.1 (C-4); 160.3 (COPh); 166.5 (C-2).

**2 $\beta$  isomer:**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 3.13-3.30 (m, 1H, H-2'a or H-2'b); 3.54-3.71 (m, 1H, H-2'a or H-2'b); 4.41-4.60 (m, 1H, H-5'a or H-5'b); 4.70-4.77 (m, 1H, H-5'a or H-5'b); 5.90 (pseudo d, 1H, H-4'); 6.55 (dd, 1H, H-1'); 7.47-8.15 (m, 6H, H arom. and H-6); 9.29 (s br, 1H, NH).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$ : 37.9 (C-2'); 64.9 (C-5); 68.7 (C-5'); 83.6 (C-4'); 86.9 (C-1'); 129.0-134.0 (C arom.); 144.3 (C-6); 150.1 (C-4); 160.1 (COPh); 166.4 (C-2). Anal. ( $\text{C}_{15}\text{H}_{13}\text{N}_2\text{O}_5\text{SI}$ ) C, H, N. MS (FAB+)  $m/z$  461  $\text{MH}^+$ .

#### 2',3'-dideoxy-5-iodo-3'-thiauridine **4 $\alpha$** and **4 $\beta$** :

Compounds **2 $\alpha$**  and **2 $\beta$**  (0.70 g) were dissolved in a solution of  $\text{NH}_3/\text{MeOH}$  (12 mL) and stirred overnight at room temperature. After solvent evaporation, the two isomers were separated over silica gel by flash column chromatography using  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  (95:5) as eluent. It gave 0.405 g of compounds **4 $\alpha$**  and **4 $\beta$**  as white solid. Yield (75%). TLC ( $\text{CH}_2\text{Cl}_2/\text{MeOH}$  95:5)  $R_f$  = 0.21. mp = 193-195°C for **4 $\alpha$**  and mp = 169-170°C for **4 $\beta$** .

**4 $\alpha$  isomer:**  $^1\text{H}$  NMR ( $\text{CD}_3\text{COCD}_3$ )  $\delta$ : 3.33-3.44 (m, 2H, 2H-2'); 3.51-3.73 (m, 2H, 2H-5'); 4.62 (t, 1H,  $J$  = 5.66 Hz, OH); 5.33 (t, 1H,  $J$  = 3.54 Hz, H-4'); 6.27 (dd, 1H, H-1'); 8.06 (s, 1H, H-6); 10.43 (s br, 1H, NH).

**4 $\beta$  isomer:**  $^1\text{H}$  NMR ( $\text{CD}_3\text{COCD}_3$ )  $\delta$ : 3.31-3.53 (m, 2H, 2H-2'); 3.57-3.62 (m, 2H, 2H-5'); 4.28 (t, 1H,  $J$  = 6.22 Hz, OH); 5.71 (t, 1H,  $J$  = 4.81 Hz, H-4'); 6.46 (dd, 1H, H-1'); 8.63 (s, 1H, H-6); 10.41 (s br, 1H, NH). Anal. ( $\text{C}_8\text{H}_9\text{N}_2\text{O}_4\text{SI}$ ) C, H, N. MS (FAB+)  $m/z$  357  $\text{MH}^+$ .

#### 2',3'-dideoxy-5'-benzoyloxymethyl-3'-thiauridine- 5-carboxylic acid **6 $\alpha$** and **6 $\beta$** :

A suspension of 5-carboxyuracil (1.3 eq, 0.5 g, 3.2 mmol) and trimethylsilyl chloride (13eq, 2.7 mL, 32.0 mmol) containing a catalytic amount of  $(\text{NH}_4)_2\text{SO}_4$  in 1,1,1,3,3,3-hexamethyldisilazane (HMDS, 10 mL) was refluxed under nitrogen until a clear solution was obtained (3h). The solution was allowed to cool to room temperature and the HMDS removed under reduced pressure. The resulting solid was resuspended in anhydrous 1,2-dichloromethane (5 mL) and **1** (1.0 eq, 0.7 g, 2.5 mmol) was added. This suspension was treated with trimethylsilyl triflate (0.5 mL). The reaction mixture which became gradually a clear solution was allowed to stir for 8h. It was then evaporated under reduced pressure. An aqueous solution (5%)  $\text{NaHCO}_3$  was added and the product extracted with ethyl acetate. The

organic phases were washed with H<sub>2</sub>O and dried over Na<sub>2</sub>SO<sub>4</sub> to give after evaporation 0.83 g of white solid which was recrystallized in CH<sub>2</sub>Cl<sub>2</sub>. Yield (90%). TLC (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 9:1) R<sub>f</sub> = 0.15.

**6 $\alpha$  isomer:** <sup>1</sup>H NMR (DMSO d<sub>6</sub>)  $\delta$  : 3.30-3.60 (m, 2H, 2H-2'), 4.20-4.30 (m, 1H, Ha-5' or Hb-5'), 4.50-4.60 (m, 1H, Ha-5' or Hb-5'), 5.50 (t, 1H, J = 3.78 Hz, H-4'), 6.20 (pseudo t, 1H, H-1'), 7.40-7.90 (m, 5H, H arom.), 8.30 (s, 1H, H-6), 12.00 (br s, 1H, COOH)

**6 $\beta$  isomer:** <sup>1</sup>H NMR (DMSO d<sub>6</sub>)  $\delta$  : 3.30-3.60 (m, 2H, 2H-2'), 4.20-4.30 (m, 1H, Ha-5' or Hb-5'), 4.50-4.60 (m, 1H, Ha-5' or Hb-5'), 5.90 (pseudo t, 1H, H-4'), 6.40 (dd, 1H, H-1'), 7.40-7.90 (m, 5H, H arom.), 8.50 (s, 1H, H-6), 12.00 (br s, 1H, COOH). Anal. (C<sub>16</sub>H<sub>14</sub>N<sub>2</sub>O<sub>7</sub>S) C, H, N. MS(LS/MS)<sup>+</sup> m/z 379 MH<sup>+</sup>.

**5-(2-N-*tert*-Butoxycarbonyl)-ethylamido-2',3'-dideoxy-5'-benzoyloxymethyl-3'-thiauridine 7 $\alpha$  and 7 $\beta$ :**

The **6 $\alpha$**  and **6 $\beta$**  isomers (1 eq, 0.20 g, 0.52 mmol) were dissolved in anhydrous CH<sub>2</sub>Cl<sub>2</sub> (5 mL). DCC (1.1 eq, 0.12 g, 0.58 mmol) and HOBT (1.1 eq, 0.08 g, 0.58 mmol) were added. The mixture was allowed to stir for 4h at room temperature under nitrogen. When a precipitation occurred, aminoethylcarboxamide (1.1eq, 0.10 mL, 0.58 mmol) and N,N'-diisopropylethylenediamine (4eq, 0.48 mL, 2.40 mmol) were added. The solution was stirred for 5h, concentrated under reduced pressure, it was washed with H<sub>2</sub>O (25 mL) and extracted with ethyl acetate. The resulting organic phases were successively washed with a 5% solution of citric acid (50 mL), a 10% sodium chloride solution (50 mL), and then dry with Na<sub>2</sub>SO<sub>4</sub>. The solution was concentrated under reduced pressure. Compounds **7 $\alpha$**  and **7 $\beta$**  were obtained in a ratio 1:1 after purification by flash chromatography over silica gel using CH<sub>2</sub>Cl<sub>2</sub>/MeOH (96:4) as eluent. It gave 0.18 g of a white solid. Yield (65%). TLC (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 9:1) R<sub>f</sub> = 0.50. mp = 103-105°C.

**7 $\alpha$  isomer:** <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  : 1.25 (s, 9H, tBu), 3.20-3.70 (m, 6H, 2H-2' and CH<sub>2</sub>-CH<sub>2</sub>), 4.30-4.55 (m, 1H, Ha-5' or Hb-5'), 4.65-4.80 (m, 1H, Ha-5' or Hb-5'), 4.90 (br s, 1H, NHCO), 5.50 (t, 1H, J = 4.43 Hz, H-4'), 6.35 (t, 1H, J = 5.13 Hz, H-1'), 7.40-8.10 (m, 5H, H arom.), 8.55 (s, 1H, H-6), 9.00 (s, 1H, NH-3).

**7 $\beta$  isomer:** <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  : 1.25 (s, 9H, tBu), 3.20-3.70 (m, 6H, 2H-2' and CH<sub>2</sub>CH<sub>2</sub>), 4.30-4.55 (m, 1H, Ha-5' or Hb-5'), 4.65-4.80 (m, 1H, Ha-5' or Hb-5'), 4.90 (br s, 1H, NHCO), 5.95 (pseudo t, 1H, H-4'), 6.50 (dd, 1H, H-1'), 7.40-8.10 (m, 5H, H arom.), 8.80 (s, 1H, H-6), 9.05 (s, 1H, NH-3).

**5-(2-N-*tert*-Butoxycarbonyl)-ethylamido-2',3'-dideoxy-3'-thiauridine 8 $\alpha$  and 8 $\beta$ :**

A mixture of compounds **7 $\alpha$**  and **7 $\beta$**  (0.083 g) was dissolved in a solution of NH<sub>3</sub>/MeOH (5 mL) and stirred overnight at room temperature. After solvent evaporation the two isomers were separated over silica gel by flash column chromatography using CH<sub>2</sub>Cl<sub>2</sub>/MeOH (49:1) as eluent. It gave 0.060 g of colourless oil. Yield (87%).

**8 $\alpha$  isomer:** TLC (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 9:1) R<sub>f</sub> = 0.39. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 1.25 (s, 9H, tBu), 3.10-3.80 (m, 6H, 2H-2', 2H-5' and CH<sub>2</sub>-CH<sub>2</sub>), 5.20 (br s, 1H, NHCO), 5.50 (s, 1H, H-4'), 6.35 (pseudo t, 1H, H-1'), 8.55 (pseudo d, 1H, H-6), 9.00 (s, 1H, NH-3).

**8 $\beta$  isomer:** TLC (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 9:1) R<sub>f</sub> = 0.46. <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$ : 1.25 (s, 9H, tBu), 3.20-3.50 (m, 6H, 2H-2' and -(CH<sub>2</sub>)<sub>2</sub>-), 3.80-4.20 (m, 2H, 2H-5'), 5.15 (s br, 1H, NHCO), 5.30 (pseudo t, 1H, H-4'), 6.30 (pseudo d, 1H, H-1'), 8.80 (s, 1H, H-6), 9.05 (s, 1H, NH-3). Anal. (C<sub>16</sub>H<sub>24</sub>N<sub>4</sub>O<sub>7</sub>S<sub>1</sub>) C, H, N. MS(FAB+) m/z 417 MH<sup>+</sup>.

#### 5-(2-ethylamine)-amido-2',3'-dideoxy-3'-thiauridine **9 $\alpha$** and **9 $\beta$** :

Compounds **8 $\alpha$**  (0.018 g) and **8 $\beta$**  (0.020 g) were separately dissolved in CH<sub>2</sub>Cl<sub>2</sub> (5 mL) with an excess of trifluoroacetic acid (2 mL). The solution was stirred for 1h under nitrogen until the starting material disappeared. It was then concentrated under reduced pressure to give a brown oil which was triturated with dichloromethane/hexane to give compounds **9 $\alpha$**  (0.008 g), 68% and **9 $\beta$**  (0.015 g), 98%.

**9 $\alpha$  isomer:** TLC (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 9:4) R<sub>f</sub> = 0.38. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$ : 2.85 (s br, 2H, NH<sub>2</sub>), 3.15-3.60 (m, 8H, 2H-2', 2H-5' and CH<sub>2</sub>-CH<sub>2</sub>), 5.15 (br s, 1H, NHCO), 5.40 (t, 1H, J = 4.82 Hz, H-4'), 6.34 (dd, 1H, H-1'), 8.30 (s, 1H, H-6), 8.76 (t, 1H, J = 5.76 Hz, NH-3), 11.94 (s, 1H, OH).

**9 $\beta$  isomer:** TLC (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 9:4) R<sub>f</sub> = 0.50. <sup>1</sup>H NMR (DMSO-d<sub>6</sub>)  $\delta$ : 2.80 (s br, 2H, NH<sub>2</sub>), 3.20-3.80 (m, 8H, 2H-2', 2H-5' and -(CH<sub>2</sub>)<sub>2</sub>-), 5.14 (t, 1H, J = 4.95 Hz, H-4'), 5.31 (br s, 1H, NHCO), 6.16 (pseudo d, 1H, H-1'), 8.58 (s, 1H, H-6), 8.75 (t, 1H, J = 5.85 Hz, NH-3), 11.92 (s, 1H, OH). Anal. (C<sub>11</sub>H<sub>16</sub>N<sub>4</sub>O<sub>5</sub>S) C, H, N. MS(FAB+) m/z 317 MH<sup>+</sup>.

#### Virology

The representative compounds were tested *in vitro* for their abilities to inhibit HIV-1 infection in MT<sub>4</sub> cells culture. The fusogenic effect of HIV in the MT<sub>4</sub> cell line was determined as described by Rey *et al* (21,22). A total of 3 $\times$ 10<sup>5</sup> MT<sub>4</sub> cells were infected with 100  $\mu$ l of diluted virus for 1 h at 37°C. After three washes, the infected cells were cultured in 24-well cell culture plates in the presence of the inhibitor. The appearance of syncytia was measured with an inverted optical microscope, 5 days after infection. The inhibitory concentration was expressed as the concentration of the tested compound which causes 50% inhibition of syncytia formation (IC<sub>50</sub>) but was not toxic for the cells. For toxicity testing, three replication cultures of each uninfected MT<sub>4</sub> cells (2 $\times$ 10<sup>5</sup> cells) were incubated with various concentrations of 2',3'-dideoxy-3'-thiacytidine analogues. Cell viability was determined 6 days from drug addition by trypan blue exclusion.

#### ACKNOWLEDGEMENTS

We are very grateful to Dr. G. Pepe and M. Meyer (C.R.M.C.2-CNRS, Luminy, France) for their help in Molecular Modélisation. We thank Dr. M. Noailly (Faculté de Pharmacie,

Université Aix-Marseille II) for the determination of NMR data. We are indebted to E. Abdili for technical assistance in antiviral activity evaluation. This research was financially supported by the Institut National de la Santé et de la Recherche Médicale (INSERM) and by the Agence Nationale pour la Valorisation de la Recherche (ANVAR).

## REFERENCES

- 1 E. De Clercq. *J. Med. Chem.* **1995**, 38, 2481.
- 2 P. Van Roey, E. Taylor, C. Chu and R. Schinazi. *Ann. New York Acad. Sci.* **1991**, 617, 29.
- 3 E. Scrocco and J. Tomasi. *Adv. Quant. Chem.* **1978**, 11, 115.
- 4 G. Pepe. *J. Mol. Graphics.* **1989**, 7, 233.
- 5 G. Pepe and D. Siri. (Ed. J. L. Rivail). Elsevier. Science, Amsterdam. **1990**, 93.
- 6 M. Camplo, V. Niddam, P. Barthelemy, P. Faury, N. Mourier, V. Simon, A-S. Charvet, C. Trabaud, J-C. Graciet, J-C. Chermann and J-L. Kraus. *Eur. J. Med. Chem.* **1995**, 30, 789.
- 7 N. Mourier, C. Trabaud, J-C. Graciet, V. Simon, V. Niddam, P. Faury, A-S. Charvet, M. Camplo, J-C. Chermann and J-L. Kraus. *Nucleosides and Nucleotides.* **1995**, 14, 1393.
- 8 B. Belleau, N. Nguyen-Ba. US. PATENT N° 5,047,407. **1989**.
- 9 M. St Clair, C.A. Richards, T. Spector, K. Weinhold, W. Miller, A. Langlois and P. Furman. *Antimicrob. Agents. Chemother.* **1987**, 31, 1972.
- 10 P. Furman, J. Fyfe, M. St Clair, K. Weinhold, D.C. Rideout, G. Freeman, S. Nusinoff-Lehrman, D. Bolognesi, S. Broder and D. Barry. *Proc. Natl. Acad. Sci. USA.* **1986**, 83, 8333.
- 11 M. Greenberg, H. Allaudeen and M. Hershfield. *Ann. N.Y. Acad. Sci.* **1990**, 616, 517.
- 12 D. Humber, M. Jones, J. Payne, M. Ramsay, B. Zacharie, H. Jiu, A. Siddiqui, C. Evans, A. Tse and T. Mansour. *Tetrahedron Lett.* **1992**, 33, 4625.
- 13 J. Beach, L. Jeong, A. Alves, D. Pohl, H. Kim, C. Chang, S. Doong, R. Schinazi, Y. Cheng and C. Chu. *J. Org. Chem.* **1992**, 57, 2217.
- 14 J-L. Kraus and G. Attardo. *Chirality.* **1993**, 5, 97.
- 15 P-G. Mattingly. *Synthesis.* **1990**, 366.
- 16 B. Castro, J. Dormoy, B. Dourtoglou, G. Evin, C. Selve and J. Ziegler. *Synthesis.* **1976**, 751.
- 17 M. Mikolasczyk and P. Kielbasinski. *Tetrahedron.* **1981**, 37, 233.
- 18 D. Nitecki, B. Halpern and S. Westley. *J. Org. Chem.* **1968**, 33, 864.
- 19 B. Halpern and D. Nitecki. *Tetrahedron Lett.* **1967**, 7, 3031.
- 20 F. Rey, F. Barré-Sinoussi, H. Schindtmayerova, J-C. Chermann. *J. Virol. Methods.* **1987**, 16, 239.

- 21 F. Rey, G. Donker, I. Hirsch, J-C. Chermann. *Virology*. **1991**, *181*, 165.
- 22 S. Harada, Y. Koyanagi, N. Yamamoto. *Science*. **1985**, *229*, 563.

Received November 10, 1995

Accepted April 10, 1996